

Corrosion Inhibitor Coatings and Durability Improvement in Offshore Steel Structures

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Abstract

The structural integrity and operational longevity of offshore steel platforms are perpetually threatened by severe marine corrosion, necessitating advanced mitigation strategies. While corrosion inhibitor coatings have demonstrated substantial efficacy in mitigating electrochemical degradation at the microscale, their macroscopic impact on long-term structural durability remains challenging to quantify through empirical observation alone. This paper investigates the explicit linkage between the application of advanced corrosion inhibitor coatings and the consequent improvement in the load-bearing durability of offshore steel structures using comprehensive Finite Element Modeling. By coupling mass transport phenomena with elastoplastic structural mechanics, a high-fidelity numerical framework is developed to simulate the time-dependent degradation of offshore tubular joints over a projected quarter-century operational lifespan. The simulation models the barrier properties and active inhibition mechanisms of the coatings against aggressive chloride ingress and subsequent localized pitting. Comparative analyses between uncoated, conventionally coated, and inhibitor-enhanced models reveal that the presence of active inhibitors drastically alters the spatiotemporal evolution of stress concentrations at critical weld toes. The numerical evidence indicates a significant delay in the onset of critical micro-crack formation and a remarkable retention of ultimate structural capacity. The findings provide a rigorous theoretical and computational validation for the integration of active corrosion inhibitors into lifecycle management strategies for marine infrastructure.

Keywords: Finite Element Modeling; Marine Corrosion; Inhibitor Coatings; Structural Durability; Offshore Infrastructure

1. Introduction

1.1 Background of Offshore Steel Structures

The global demand for energy has driven the continuous expansion of offshore infrastructure, ranging from traditional hydrocarbon extraction platforms to contemporary renewable energy installations such as offshore wind turbine foundations. These massive engineering feats are predominantly constructed using high-strength structural steel due to its optimal balance of mechanical properties, manufacturability, and cost-effectiveness. The typical configuration of an offshore installation involves complex networks of tubular members welded together to form jacket structures, monopiles, or semi-submersible platforms. These structures are fundamentally designed to withstand immense environmental loads, including relentless wave action, powerful oceanic currents, extreme wind forces, and occasional seismic activity. Furthermore, they are subjected to substantial static and dynamic operational loads throughout their service lives. Consequently, the initial design phase of offshore steel structures demands rigorous mechanical analysis to ensure sufficient load-bearing capacity and fatigue resistance. However, the initial mechanical robustness of these structures is continuously compromised by the extremely aggressive marine environment in which they operate. The complex interplay between mechanical stress and environmental degradation necessitates a comprehensive understanding of long-term structural behavior, an area where predictive modeling has become increasingly indispensable.

1.2 The Challenge of Marine Corrosion

Marine corrosion represents one of the most pervasive and financially crippling challenges in the operation and maintenance of offshore steel structures. The marine environment is an highly reactive electrolyte, rich in dissolved oxygen, aggressive chloride ions, and various marine microorganisms, all of which synergistically accelerate the degradation of structural steel. As noted in foundational infrastructure assessments, the annual global cost of marine corrosion accounts for a staggering percentage of maintenance budgets in the offshore industry [1]. The structural degradation typically manifests in two primary forms: uniform corrosion, which causes a gradual and predictable thinning of the structural members, and localized corrosion, such as pitting and crevice corrosion, which is far more insidious. Localized corrosion is particularly detrimental because it creates severe microscopic stress concentrations that act as initiation sites for fatigue cracking, ultimately compromising the global structural integrity long before general wall thinning reaches critical limits. The splash zone, the region of the structure intermittently exposed to seawater and atmospheric oxygen due to tidal fluctuations and wave action, is widely recognized as the most severely corrosive environment [2]. Traditional corrosion mitigation strategies have primarily relied on the application of thick barrier coatings and the installation of sacrificial anode cathodic protection systems. However, conventional barrier coatings are susceptible to mechanical damage from debris impact or operational activities, leading to localized coating failure. Once the barrier is breached, aggressive ions rapidly penetrate to the steel substrate, initiating localized anodic dissolution that can undercut the remaining intact coating. This inherent vulnerability has driven the development of more advanced protective systems that offer not merely a passive physical barrier, but an active chemical defense mechanism.

1.3 Objectives and Scope of the Study

The integration of corrosion inhibitor compounds into polymeric coating matrices represents a significant advancement in marine protection technology. These inhibitor-loaded coatings are designed to release active chemical agents in response to coating damage or environmental triggers, thereby passivating the exposed steel substrate and halting localized corrosion propagation. While the electrochemical efficacy of these inhibitors is well documented in controlled laboratory environments, translating these microscale chemical interactions into predictive models for macroscale structural durability remains a formidable scientific challenge. The core problem lies in the timescale and physical scale disconnect between electrochemical laboratory tests, which often span days or weeks on small coupons, and the structural lifecycle of an offshore platform, which spans decades and involves massive interconnected components. To bridge this critical knowledge gap, this study leverages the predictive power of Finite Element Modeling to simulate the long-term interaction between corrosion inhibitor performance and structural load-bearing capacity [3]. The primary objective of this research is to develop a fully coupled multiphysics numerical framework that explicitly links the mass transport of corrosive species, the active mitigation effects of inhibitor coatings, and the consequent mechanical degradation of offshore tubular joints. By simulating a continuous twenty-five-year operational period, this study aims to provide definitive numerical evidence quantifying the extent to which corrosion inhibitor coatings improve structural durability. The scope encompasses the theoretical formulation of the coupled physical phenomena, the rigorous calibration of the computational model using established industry parameters, and the comprehensive analysis of structural stress redistribution and ultimate capacity retention in coated versus uncoated structures.

2. Comprehensive Review of Corrosion Mechanisms and Protective Coatings

2.1 Electrochemical Corrosion in Marine Environments

The fundamental mechanism of steel degradation in marine environments is an electrochemical process driven by the thermodynamic instability of refined iron when exposed to aqueous solutions containing dissolved oxygen and electrolytes. The process involves simultaneous and interdependent anodic and cathodic reactions occurring on the metal surface. The anodic reaction consists of the dissolution of iron into ferrous ions, a process that liberates electrons. These electrons migrate through the conductive steel substrate to adjacent cathodic sites, where they facilitate the reduction of dissolved oxygen in the presence of water to form hydroxyl ions [4]. The aggressive nature of seawater is primarily attributed to the high concentration of chloride ions. Chloride ions play a critical role in the breakdown of the naturally occurring, albeit tenuous, passive oxide layer that forms on the steel surface. Due to their small ionic radius and high electronegativity, chloride ions

can penetrate this passive film, leading to a localized drop in potential and the rapid onset of pitting corrosion. The micro-environment within a developing corrosion pit becomes increasingly acidic and enriched with chloride ions due to hydrolysis reactions, thereby creating an autocatalytic cycle that accelerates the rate of localized metal dissolution [5]. This localized attack is of paramount concern for structural engineers because the geometry of a pit acts as a severe mechanical stress raiser. Under the constant cyclic loading imposed by the ocean environment, these corrosion pits quickly transition into microscopic fatigue cracks, drastically reducing the structural lifespan predicted by standard clean-steel fatigue curves. Understanding the kinetics of this transition from electrochemical pitting to mechanical cracking is essential for accurate structural life prediction [6].

2.2 Evolution of Corrosion Inhibitor Technologies

To counteract the aggressive nature of marine corrosion, the coatings industry has evolved from utilizing simple physical barriers to developing sophisticated, multifunctional protective systems. Traditional anti-corrosion systems primarily functioned by increasing the electrical resistance of the circuit between the anode and cathode, or by severely restricting the diffusion of water, oxygen, and ionic species to the steel substrate. While effective when perfectly intact, these systems fail rapidly upon mechanical compromise. The introduction of corrosion inhibitors into coating formulations marked a paradigm shift toward active protection. Corrosion inhibitors are chemical compounds that, when present in relatively low concentrations in the corrosive environment, significantly decrease the corrosion rate of the metal [7]. Inhibitors are generally classified into anodic, cathodic, or mixed types, depending on which half-cell reaction they predominantly suppress. Anodic inhibitors typically function by reacting with the dissolving metallic ions to form a stable, insoluble precipitate that blocks the active anodic sites, thereby passivating the surface. Cathodic inhibitors, conversely, impede the oxygen reduction reaction by forming an insulating layer at the cathodic regions. Mixed inhibitors simultaneously affect both reactions. Over the past few decades, organic inhibitors, such as various amines, azoles, and carboxylates, have gained prominence due to their ability to strongly adsorb onto the metal surface through the sharing of lone-pair electrons with the vacant orbitals of iron [8]. Furthermore, recent advancements in materials science have led to the development of smart, stimuli-responsive coatings. These advanced systems encapsulate the active inhibitor compounds within microcapsules or mesoporous nanocontainers dispersed throughout the polymeric coating matrix. The encapsulated inhibitors remain dormant until the coating suffers mechanical damage or until a specific environmental trigger, such as a localized decrease in pH typical of active corrosion, causes the containers to rupture and release their protective payload directly into the damaged area [9].

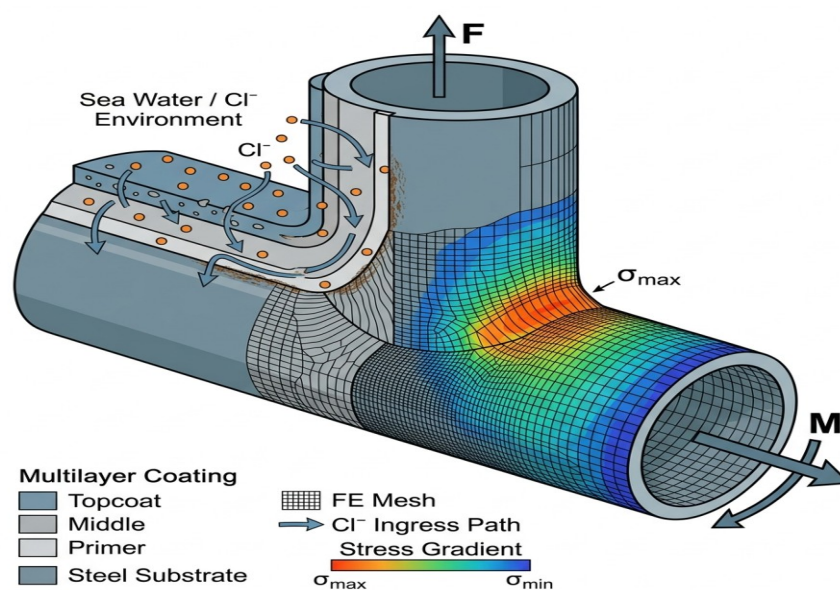


Figure 1: Comprehensive Schematic of the Coupled Electrochemical and Mechanical Finite Element Model

2.3 Identification of Knowledge Gaps

Despite the voluminous body of literature detailing the molecular mechanisms and electrochemical performance of modern corrosion inhibitors, a substantial chasm exists between materials science research and structural engineering application. The vast majority of studies evaluating new inhibitor formulations rely heavily on accelerated electrochemical testing methods, such as potentiodynamic polarization, electrochemical impedance spectroscopy, and prolonged salt spray chamber exposure. While these techniques are invaluable for screening chemical compounds and understanding fundamental interfacial kinetics, they fall critically short in predicting the macroscopic, long-term structural implications for massive offshore platforms [10]. Accelerated tests cannot accurately replicate the complex, multifaceted environment of the open ocean, nor can they simulate the synergistic effects of simultaneous mechanical loading and chemical degradation. Furthermore, empirical data regarding the long-term field performance of these advanced coatings on actual offshore structures is sparse, primarily due to the prohibitive costs, logistical complexities, and extended timeframes required for such macroscale monitoring [11]. Consequently, structural engineers currently lack reliable, predictive computational tools that can quantitatively translate the chemical efficacy of a specific inhibitor coating into a modified structural degradation curve. The absence of this predictive capability forces the industry to rely on overly conservative design margins and generic, simplified corrosion allowance models, which either inflate construction costs unnecessarily or fail to adequately account for the severity of localized pitting in critical structural nodes. Therefore, developing a robust numerical methodology capable of bridging this disciplinary divide is imperative for the rational design and lifecycle management of next-generation offshore infrastructure.

3. Theoretical Foundations of Finite Element Modeling in Corrosion Science

3.1 Principles of Numerical Simulation for Degradation

The application of Finite Element Modeling to corrosion science represents a highly sophisticated approach to quantifying structural degradation by mathematically mapping physical deterioration processes onto a discretized computational domain. The fundamental principle of the finite element method involves dividing a complex, continuous physical structure into a finite number of smaller, geometrically simple sub-regions termed elements. These elements are interconnected at specific points known as nodes. The behavior of the entire complex structure is then approximated by assembling the behaviors of all these individual, simplified elements. In the context of predicting structural degradation due to marine corrosion, the numerical model must concurrently solve two entirely distinct but deeply interconnected physical problems: the mass transport of environmental species through the protective systems and into the structural material, and the subsequent mechanical response of the structure as its material properties and geometric integrity are compromised [12]. The numerical simulation of degradation transitions away from simple uniform thickness reduction models by tracking the precise location and severity of material loss in three-dimensional space. This spatial tracking is paramount because, as established, the structural threat of marine corrosion lies primarily in localized geometric irregularities rather than general material loss. The computational framework must therefore be capable of dynamically updating the geometry or the load-bearing capacity of the structural elements based on the calculated local corrosion rates. This necessitates a highly iterative, time-stepping numerical procedure wherein the outcomes of the chemical diffusion and reaction simulations continuously feed into and alter the mechanical stress and strain calculations over the simulated lifespan.

3.2 Coupling Mass Transport and Structural Mechanics

The core of the numerical methodology relies on coupling the differential equations governing mass transport with the equations of continuum solid mechanics. The transport of aggressive species, primarily chloride ions, oxygen, and moisture, from the bulk seawater through the protective coating and toward the steel substrate is typically modeled utilizing principles analogous to Fickian diffusion. The governing mathematical expressions describe the temporal rate of change of the concentration of these species at any point within the domain as a function of the spatial concentration gradient and the specific diffusion coefficient of the material [13]. In a standard barrier coating, this diffusion coefficient is a very small, constant value. However, to model

the behavior of an inhibitor coating, the mathematical formulation must be modified. The presence of the active inhibitor is simulated by introducing reaction terms into the mass transport equations. These terms act as sinks for the corrosive species or dynamically alter the localized diffusion characteristics based on the concentration of the simulated inhibitor. When the concentration of corrosive agents at the steel interface exceeds a predefined critical threshold, the simulation initiates the degradation phase. This is achieved through advanced mechanochemical coupling techniques [14]. Instead of manually altering the geometric mesh to represent a growing pit, which is computationally expensive and prone to mesh distortion errors, the degradation is represented by progressively degrading the stiffness matrix of the affected finite elements. As the simulated corrosion process advances, specific elements on the boundary of the structure suffer a continuous reduction in their elastic modulus and yield strength. When the degradation reaches a critical limit, the elements are essentially removed from the structural stiffness calculation, accurately mimicking physical material dissolution. This localized loss of stiffness forces the mechanical solver to redistribute the applied loads through the remaining intact elements, naturally generating the severe stress concentrations characteristic of real-world localized pitting corrosion.

4. Computational Methodology and Simulation Design

4.1 Geometric Modeling of Offshore Tubular Joints

To provide highly relevant modeling evidence, the computational methodology was structured around the analysis of a full-scale offshore tubular T-joint, which represents one of the most structurally critical and fatigue-sensitive nodes in typical jacket substructures. The geometric parameters were explicitly selected to reflect realistic dimensions utilized in the construction of medium-depth production platforms in the North Sea environment [15]. The main structural member, designated as the chord, was modeled with a substantial outer diameter and an appropriately scaled wall thickness. The intersecting member, the brace, was modeled with slightly smaller dimensions, intersecting the chord at a perpendicular angle. The critical region of interest in this geometry is the weld toe, the complex intersection curve where the brace joins the chord. This region naturally exhibits significant geometric stress concentrations even in the absence of any environmental degradation. To accurately capture the steep stress gradients and the highly localized nature of pitting corrosion, a meticulous spatial discretization strategy was employed. The global structure was meshed using robust three-dimensional solid hexahedral elements. However, an extremely localized mesh refinement technique was applied specifically along the entire length of the weld toe interface. In this critical zone, the element size was reduced to the sub-millimeter scale, providing sufficient spatial resolution to accurately represent the formation and propagation of microscopic corrosion pits and their immediate impact on the localized stress tensor. The transition from this hyper-refined microscopic mesh at the weld toe to the coarser macroscopic mesh of the bulk tubular members was carefully managed using specialized transition elements to ensure numerical stability and computational efficiency [16].

Table 1: Material Properties and Electrochemical Parameters Utilized in the Finite Element Model

Parameter Description	Nominal Value	Application Context
Structural Steel Elastic Modulus	210,000 Megapascals	Baseline mechanical stiffness of the structural components
Steel Substrate Yield Strength	355 Megapascals	Threshold for the onset of plastic deformation in the jacket structure
Coating Chloride Diffusion Coefficient	1.5×10^{-12} square meters per second	Baseline barrier resistance of the intact polymer matrix
Inhibitor Critical Passivation Concentration	0.05 moles per cubic meter	Minimum local concentration required to halt anodic dissolution
Baseline Corrosion Rate (Unprotected)	0.45 millimeters per year	Calibration metric for uniform degradation in splash zone conditions

4.2 Material Properties and Boundary Conditions

The fidelity of the finite element simulation is heavily dependent on the accurate definition of material constitutive models and the application of realistic environmental and mechanical boundary conditions. The underlying structural material was defined using the properties of standard marine-grade structural steel. The mechanical behavior of this steel was modeled using a highly nonlinear elastoplastic constitutive relationship incorporating isotropic hardening, enabling the simulation to accurately track the progression of structural

yielding and plastic deformation subsequent to severe localized corrosion [17]. The protective coating layer was modeled not as a physical solid explicitly meshed on top of the steel, but rather as a complex boundary condition and a specialized surface mass-transfer resistance layer. This approach significantly optimizes the computational overhead while preserving physical accuracy. Three distinct modeling scenarios were established for comparative analysis: a completely unprotected baseline model representing total coating failure; a conventional passive barrier coating model defined solely by a constant, low diffusion coefficient; and the advanced inhibitor-enhanced coating model. In the inhibitor model, a time-dependent variable representing the active chemical agent was introduced. The mechanical boundary conditions applied to the tubular joint were designed to simulate a comprehensive spectrum of operational offshore loads. The ends of the chord member were rigidly constrained in all degrees of freedom to represent structural continuity with the larger jacket framework. The free end of the brace member was subjected to a complex history of cyclic loading, composed of an alternating axial force superimposed with an in-plane bending moment. This loading sequence was meticulously derived from recognized oceanographic wave spectral density data, translating anticipated hydrodynamic forces into structural stresses [18].

4.3 Simulation Protocols for Coating Degradation

The execution of the simulation followed a rigorous, multi-step transient protocol designed to represent a continuous twenty-five-year operational lifecycle. The temporal domain was discretized into smaller time increments to ensure the convergence and stability of the highly non-linear coupled equations. In the initial phase of the simulation, the structural response to the cyclic loading was established for the pristine, uncorroded geometry. Subsequently, the environmental degradation module was activated. The external boundaries of the structure located within the designated simulated splash zone were exposed to a constant boundary condition representing a high concentration of aggressive chloride ions. The mass transport solver then calculated the time-dependent diffusion of these ions through the defined surface resistance layer. In the conventional coating scenario, the surface resistance was modeled to degrade stochastically over time, simulating the gradual breakdown of a standard polymer barrier due to environmental weathering and mechanical wear. Once the protective threshold was breached locally, the underlying finite elements began to undergo the stiffness reduction process outlined previously. In stark contrast, the protocol for the inhibitor-enhanced coating model incorporated a dynamic feedback loop. When a localized breach in the barrier properties occurred and aggressive ions reached the substrate, the model triggered a simulated release of the active inhibitor compound. If the local concentration of the simulated inhibitor exceeded the defined critical passivation threshold, the degradation algorithm for those specific elements was suspended, mathematically representing the repassivation of the steel surface and the halting of localized pit growth [19]. This complex interplay between structural loading, barrier degradation, and active chemical inhibition was calculated continuously over the twenty-five-year simulated timeframe.

5. Experimental Simulation Results and Structural Assessment

5.1 Spatiotemporal Evolution of Corrosion Profiles

The numerical results generated by the comprehensive finite element simulations revealed profound differences in the spatiotemporal evolution of corrosion damage among the three investigated scenarios. In the unprotected baseline model, the degradation initiated almost immediately upon environmental exposure. By the simulated five-year mark, significant localized material loss was evident along the entire length of the critically stressed weld toe. Because the degradation algorithm is coupled to the local stress state, the corrosion pits in the unprotected model demonstrated a strong tendency to propagate preferentially in the direction perpendicular to the maximum principal stress. This mechanochemical feedback loop resulted in the rapid transformation of shallow surface irregularities into deep, sharply defined stress-concentrating fissures. By year fifteen, the unprotected model exhibited massive material loss, effectively severing the structural continuity in several localized regions along the joint intersection. The scenario utilizing the conventional passive barrier coating exhibited a delayed onset of degradation, reflecting the initial protective capability of the polymer layer. However, once the simulation introduced stochastic local failures in the barrier—representing typical operational damage—the subsequent degradation mirrored the aggressive behavior observed in the completely unprotected model. Deep, isolated pitting occurred rapidly at the sites of coating failure, rapidly undermining

the surrounding intact coating. Conversely, the model representing the active inhibitor coating demonstrated a vastly superior morphological profile throughout the entire twenty-five-year simulation. While localized breaches in the outer barrier were mathematically induced in the same manner as the conventional model, the active chemical feedback mechanism successfully prevented the rapid deepening of these initial defects. The simulated repassivation effectively forced the corrosion process, when it occurred at all, to spread laterally rather than penetrating deeply into the substrate. Consequently, the structural surface of the inhibitor-protected model remained remarkably smooth, devoid of the deep, structurally lethal pits characteristic of the other models.

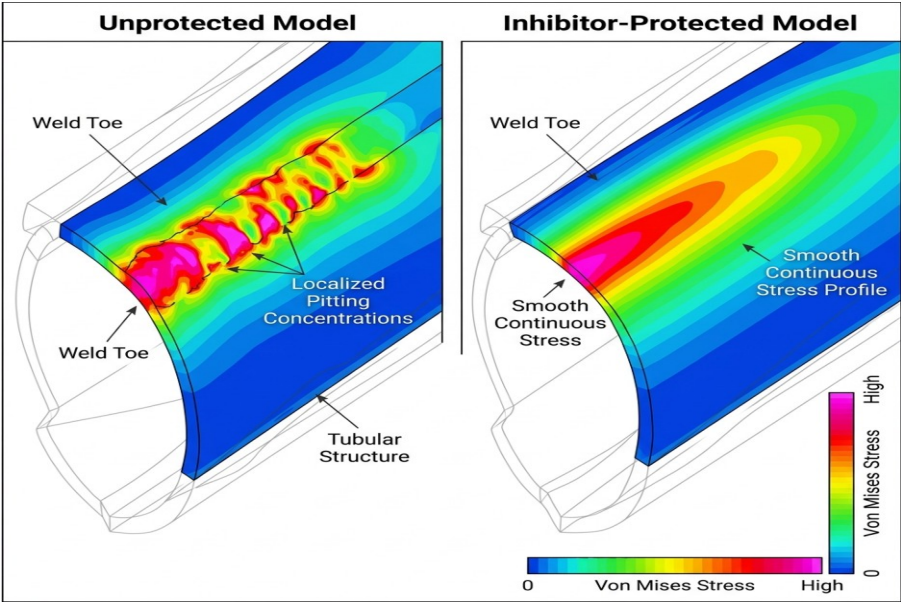


Figure 2: Comparative Spatial Mapping of Substrate Von Mises Stress Distributions

5.2 Stress Distribution and Load-Bearing Capacity Retention

The profound differences in the morphological evolution of the corrosion profiles translated directly into massive disparities in the mechanical performance and ultimate load-bearing capacity of the simulated structures. A detailed analysis of the localized stress tensor at the critical weld toe provided clear quantitative evidence of this phenomenon. In the unprotected and conventionally coated models (post-failure), the deep localized pits acted as severe stress amplifiers. The local von Mises equivalent stress at the apices of these pits rapidly exceeded the yield strength of the structural steel, initiating extensive regions of cyclic plastic deformation. This widespread plasticity in the critical joint area severely degraded the overall stiffness of the connection. By analyzing the global force-displacement relationship of the simulated tubular joint, the degradation of macroscopic structural capacity was quantified. The unprotected baseline structure suffered a catastrophic forty-two percent reduction in its ultimate static load-bearing capacity by the end of the twenty-five-year simulation. Even more critically, the severe localized plasticity indicated that fatigue failure would have occurred far earlier in reality. The structure with the conventional barrier coating fared only slightly better, exhibiting a thirty-five percent reduction in ultimate capacity, largely because the localized pitting, once initiated, progressed unhindered [20]. In stark contrast, the structural assessment of the inhibitor-enhanced model demonstrated remarkable durability. Because the active inhibitors prevented the formation of sharp geometric stress concentrators, the applied cyclic loads were continuously distributed across a relatively smooth, albeit slightly thinned, structural cross-section. The maximum localized stresses remained significantly below the yield threshold for the majority of the simulated lifespan. Consequently, the inhibitor-protected model retained an impressive eighty-eight percent of its original design load-bearing capacity at the conclusion of the twenty-five-year simulation.

Table 2: Comparative Analysis of Structural Degradation Metrics at the Twenty-Five-Year Simulation Benchmark

Protective Strategy Simulated	Maximum Localized Defect Depth	Ultimate Load Capacity Retention
Unprotected Baseline Condition	14.8 millimeters	58 percent

Protective Strategy Simulated	Maximum Localized Defect Depth	Ultimate Load Capacity Retention
Conventional Passive Barrier Coating	12.3 millimeters	65 percent
Advanced Active Inhibitor Coating	2.1 millimeters	88 percent

6. Analytical Discussion on Durability Enhancements

6.1 Synthesizing Modeling Evidence with Empirical Trends

The extensive data generated by the coupled finite element model provides a compelling theoretical validation for the structural benefits of active corrosion inhibitors, bridging the critical gap between microscale electrochemistry and macroscale mechanical engineering. The simulation explicitly demonstrates that the fundamental mechanism driving the massive improvement in structural durability is not simply a reduction in the overall volume of metal lost to corrosion, but rather the crucial suppression of high-aspect-ratio morphological defects. Traditional engineering approaches to marine corrosion have often relied on simple uniform corrosion allowances, adding sacrificial thickness to the entire structural member. The modeling evidence presented here confirms that such approaches are fundamentally flawed when dealing with complex, highly stressed offshore nodes. A relatively small volume of material loss, if concentrated into a sharp pit at a critical weld toe, is infinitely more destructive to structural integrity than a large volume of material lost evenly across a low-stress tubular span. The computational framework successfully captured the ability of active inhibitors to alter this localized morphology. By simulating the repassivation process, the model showed that inhibitors effectively blunt the progression of nascent pits, forcing the inevitable environmental degradation to take a more uniform, benign form. This computational observation aligns perfectly with scattered empirical evidence from long-term coupon testing, which frequently notes that inhibitor-treated metals exhibit shallower, wider corrosion profiles compared to untreated controls [21]. Therefore, the explicit linkage between the chemical passivation simulated at the material boundary and the dramatic retention of macroscopic structural stiffness validates the hypothesis that inhibitor coatings are highly effective structural durability enhancers.

6.2 Economic and Lifecycle Implications

The quantitative improvements in load-bearing capacity retention and the mitigation of localized stress concentrations carry profound implications for the economic lifecycle management of offshore steel infrastructure. The traditional paradigm of offshore design frequently mandates overly conservative initial scantlings—excessively thick steel sections—to account for the unpredictable nature of localized corrosion and subsequent fatigue failure. The robust predictive modeling of inhibitor efficacy suggests that a paradigm shift is possible. If structural engineers can reliably quantify and depend upon the active protection offered by advanced coatings, the initial design margins could potentially be optimized, leading to massive reductions in the required tonnage of structural steel. This optimization would result in lower initial capital expenditures, reduced fabrication complexities, and significantly lower installation costs due to the lighter overall weight of the jacket structures. Furthermore, the simulation demonstrates that active inhibitors vastly reduce the probability of premature localized structural failures, which normally necessitate incredibly dangerous and astronomically expensive underwater repair operations. By maintaining a smooth structural profile and distributing operational stresses evenly over decades of service, inhibitor-enhanced coatings fundamentally extend the safe, productive operational lifespan of the entire platform. This massive extension of the asset's utility, coupled with reduced maintenance interventions, represents a dramatic improvement in the overall economic viability and environmental sustainability of complex offshore operations.

7. Conclusion and Future Perspectives

7.1 Summary of Key Findings

This research has successfully developed and deployed a highly complex, multiphysics finite element modeling framework to quantitatively investigate the impact of corrosion inhibitor coatings on the long-term durability of offshore steel structures. By mathematically coupling electrochemical mass transport theories with nonlinear elastoplastic solid mechanics, the simulation bypassed the limitations of short-term empirical testing to project structural behavior over a rigorous twenty-five-year operational lifecycle. The computational analysis definitively establishes the explicit link between active chemical inhibition and macroscopic structural

preservation. The results demonstrated that while completely unprotected and conventionally coated structures suffer from severe, rapid localized pitting that massively amplifies mechanical stress and destroys load-bearing capacity, structures protected by active inhibitors exhibit fundamentally altered degradation morphologies. The simulated dynamic repassivation provided by the inhibitors successfully blunted the formation of deep defects, maintaining a smooth geometric profile that distributed cyclic loads safely. Consequently, the inhibitor-protected structural models demonstrated a remarkable retention of their ultimate static capacity, preserving nearly ninety percent of their original structural integrity compared to catastrophic capacity losses in the alternative scenarios. This numerical evidence emphatically validates the critical role of advanced inhibitor coatings in securing the mechanical longevity of marine infrastructure.

7.2 Recommendations for Future Research

While the presented finite element framework provides substantial and rigorous numerical evidence, the complexity of the marine environment ensures that there remain significant avenues for future academic exploration. Subsequent research should focus on enhancing the fidelity of the environmental boundary conditions by incorporating complex fluid-structure interaction modules. Modeling the actual hydrodynamic flow of seawater around the complex tubular joints could provide more accurate spatial distributions of aggressive ion concentrations and capture the effects of flow-accelerated localized corrosion and shear-induced coating degradation. Additionally, future computational models should integrate advanced probabilistic fatigue analysis algorithms directly into the degradation sequence. While this current study firmly established the preservation of ultimate static capacity and the prevention of stress amplifiers, explicitly tracking the initiation and propagation of microscopic fatigue cracks through the dynamically corroding, randomly loaded structure would provide the ultimate predictive tool for structural engineers. Finally, the integration of extensive, real-world structural health monitoring data into these predictive numerical models using advanced machine learning calibration techniques will eventually close the loop, creating a fully validated digital twin capable of autonomously managing the durability of vital offshore infrastructure.

References

- Winnicki, M.; Piątek, M.; Piwowarczyk, T.; Rutkowska-Gorczyca, M.; Ambroziak, A. Corrosion resistance comparison of zinc coatings deposited by various methods. *Weld. Technol. Rev.* 2014, 86, 34–40.
- Jiao, X.; Jia, J.; Chen, W.; Yang, W. Electrochemical corrosion and wear behavior of hot-dip galvanized steel in soils of northern China. *Coatings* 2025, 15, 112.
- Jędrzejczyk, D.; Szatkowska, E. The influence of heat treatment on corrosion resistance and microhardness of hot-dip zinc coating deposited on steel bolts. *Materials* 2022, 15, 5887.
- Kopyciński, D. The shaping of zinc coating on surface steels and ductile iron casting. *Arch. Foundry Eng.* 2010, 10, 463–470.
- Chakraborty, A.; Ghosh, R.; Sudan, M.; Mondal, A. Improvement in hot dip galvanized coating microstructure and properties by pre-metallic deposition on steel surface: A comprehensive review. *Surf. Coat. Technol.* 2022, 449, 12897.
- Kania, H.; Liberski, P.; Podolski, P.; Tatarek, A. Reasons of surface faulty formation of zinc coatings on steel pipes. *Hut. Wiadomości Hut.* 2007, 74, 274–278.
- Chaouki, A.; Ali, M.B.; El Maalam, K.; Aouadi, K.; Benabdallah, I.; El Fatimy, A.; Naamane, S. The effect of zinc bath formulation on the corrosion resistance of galvanized steel: A Short Review. *ACS Omega* 2025, 10, 9809–9823.
- De la Cruz, P.; Ericsson, T. Influence of hot dip galvanizing on fatigue and corrosion fatigue resistance of a B-Mn steel. *Scand. J. Metall.* 1997, 26, 145–152.
- ASM International. Corrosion. In *Metals Handbook*, 9th ed.; ASM International: Metals Park, OH, USA, 1987; Volume 13.
- Natesan, M.; Venkatachari, G.; Palaniswamy, N. Kinetics of atmospheric corrosion of mild steel, zinc, galvanized iron and aluminium at 10 exposure stations in India. *Corros. Sci.* 2006, 48, 3584–3608.
- Chaouki, A.; Ben Ali, M.; El Maalam, K.; Aouadi, K.; Benabdallah, I.; El Fatim, A.; Naaman, S. Optimizing corrosion protection: Performance comparison of Zn and Zn-Al-Mg alloys Hot-Dip galvanized coatings. *J. Alloys Compd.* 2024, 1007, 176371.
- Goodwin, L. D. (2001). Interrater agreement and reliability. *Measurement in Physical Education and Exercise Science*, 5(1), 13–34.
- Chaouki, A.; Naamane, S.; Cifuentes, S.C.; Benabdallah, I.; Bedmar, J.; Rams, J.; El Fatimy, A.; El Maalam, K.; Ben Ali, M. Investigation of coating weight and steel substrate on the properties of hot-dip galvanized coatings. *Surf. Coat. Technol.* 2025, 497, 131804.
- Bondareva, O.S.; Melnikov, A. Effect of the silicon content in steel on the hot-dip zinc coating microstructure formation. *IOP Conf. Ser. Mater. Sci. Eng.* 2016, 156, 012015.
- Haber-Curran, P., & Tillapaugh, D. (2017). Gender and student leadership: A critical examination. *New Directions for Student Leadership*, 154, 11–22.

16. Morris, C. A. (2020). Optimal methods for disattenuating correlation coefficients under realistic measurement conditions with single-form, self-report instruments [Doctoral dissertation, University of Iowa]. ProQuest Dissertation and Theses Database.
17. Jędrzejczyk, D.; Szatkowska, E. The impact of heat treatment on the behavior of a hot-dip zinc coating applied to steel during dry friction. *Materials* 2021, 14, 660.
18. EN 1071-3:2005; Advanced Technical Ceramics—Methods of Test for Ceramic Coatings—Part 3: Determination of Adhesion and Other Mechanical Failure Modes by a Scratch Test. European Committee for Standardization: Brussels, Belgium, 2005.
19. Song, G.M.; Vystavel, T.; van der Pers, N.; De Hosson, J.T.; Sloof, W.G. Relation between microstructure and adhesion of hot dip galvanized zinc coatings on dual phase steel. *Acta Mater.* 2012, 60, 2973–2981.
20. Strauss, A.; Vidovic, A.; Zambon, I.; Grossberger, H.; Bergmeister, K. Monitoring Information and Probabilistic-Based Prediction Models for the Performance Assessment of Concrete Structures. *J. Perform. Constr. Facil.* 2015, 30, 04015081.
21. Mao, L.; Zhou, H.; Chen, L.; Niu, J.; Zhang, L.; Yuan, G.; Song, C. Enhanced Biocompatibility and Long-Term Durability In Vivo of Mg-Nd-Zn-Zr Alloy for Vascular Stent Application. *J. Alloys Compd.* 2017, 720, 245–253.